

**Enzymatic Depolymerization Against Polymers Structure and Properties in the Perspective of Biochemical Recycling** 





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## Introduction

Biochemical recycling via enzymatic depolymerization is an alternative, sustainable approach to produce oligomers or monomers for new polymers or other value-added products [1], [2]. Even for biodegradable polymers, enzymatic depolymerization becomes promising since their degradation kinetics in real environmental conditions is found slow [3] – [5]. In this context, a psychrophilic esterase (MoPE) from the Antarctic bacterium Moraxella sp. was tested based on its ability to degrade via hydrolysis reactions a variety of biodegradable and nonbiodegradable, semi-crystalline polymers [1].

# **Experimental Part**



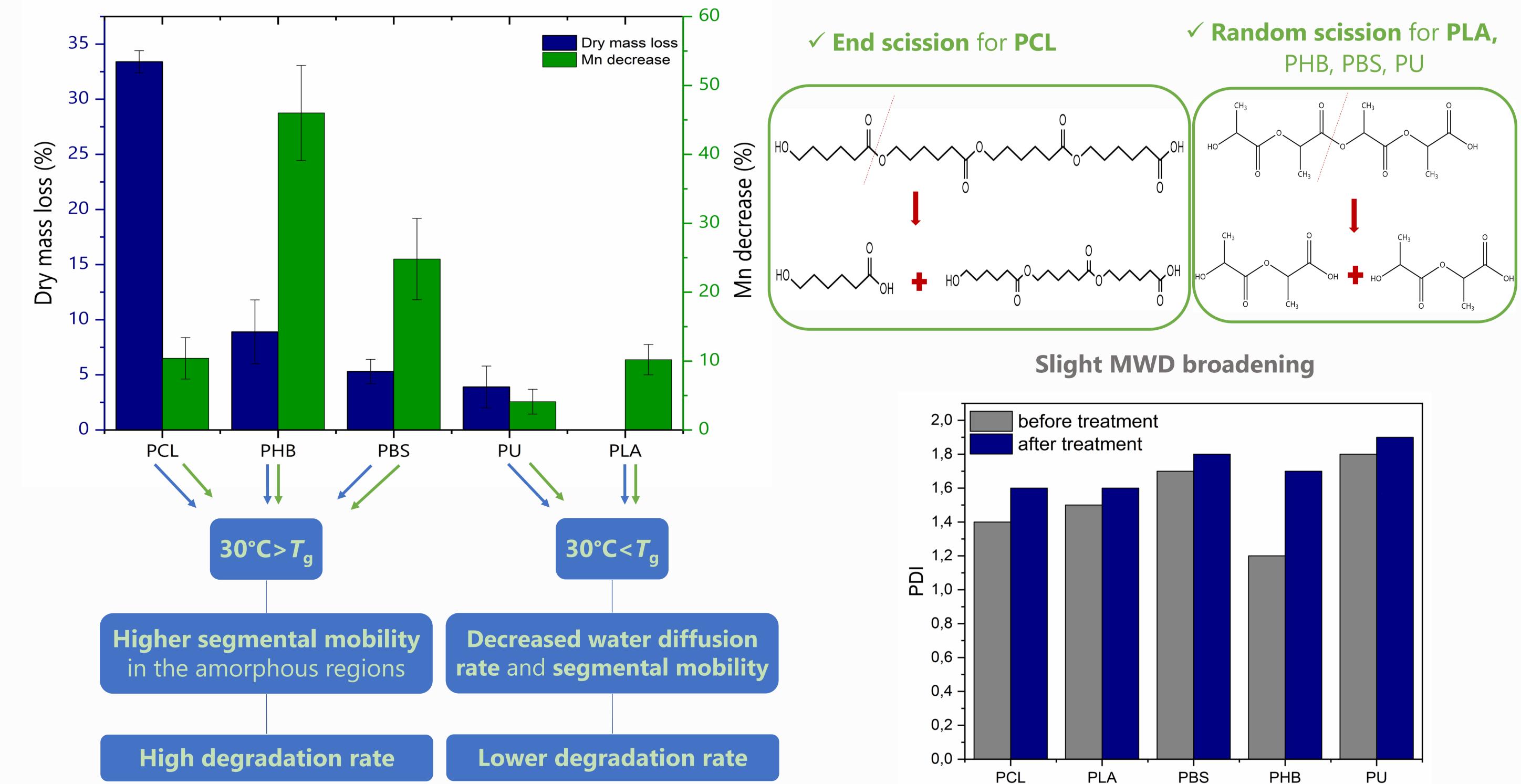
### **Results and Discussion**

**Characterization** of the target polymers and **ranking** based the most important on properties ( $T_{q_i} x_{c'} \overline{M_n}$ ) affecting enzymatic degradation.

Polymer	<b>Τ</b> <sub>g</sub> (°C )	х <sub>с</sub> (%)	M <sub>n</sub> (g/mol)
PLA	59	11	$100600 \pm 300$
PHB	-4	49	$177400 \pm 8100$
PCL	-64	45	73700± 600
PBS	-32	64	16100± 900
PU	150	_	$66500 \pm 500$

Property	Polymers Ranking								
<b>T</b> g	PCL	<	PBS	<	PHB	<	PLA	<	PU
X <sub>c</sub>	PBS	>	PHB	>	PCL	>	PLA	>	PU
Mn	PHB	>	PLA	>	PCL	>	PU	>	PBS

The enzymatic degradation efficiency was evaluated based on the determined dry mass loss, the  $M_n$  and the PDI decrease.



### PLA PBS PCL PHB

## Conclusions

MoPE was found efficient to hydrolyze biodegradable and non-biodegradable polymers with a broad range of x<sub>c</sub> from 11 up to 64%, indicating the ability of the enzyme to degrade highly crystalline polymers. The polymers' glass transition temperature was found the most crucial factor for the hydrolysis rate.

References	Acknowledgment				
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